# **Evaluation of Organic Carbon, Ash and Nitrogen Content in White Mangroves (***Laguncularia racemosa***)** from Central and Western Regions of Ghana

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### Abstract

When wood is used as fuel by burning, nitrogen dioxide  $(NO_2)$  may be formed, and NOx can enrich forest soil. However, it has a negative impact on health from the emission of high levels of NOx, including acidification of water and soils. Carbon content contributes significantly to the heating value of wood fuel, with higher carbon content giving a higher heating value. This study aimed to determine the organic carbon, nitrogen and ash content in White Mangroves (*Laguncularia Racemosa*) from Ghana's Central and Western Regions. *L. racemosa* generally showed low organic carbon, ash and nitrogen content values. The study showed 30.72 - 31.52% and 30.52 - 31.56% (organic carbon), 2.39 - 3.26% and 2.62 - 2.94% (ash), and 223.41 - 300.32 mg/Kg and 199.55 - 242.66 mg/Kg (nitrogen) in mangrove samples from Central and Western regions respectively.

#### Keywords

White Mangrove; Ash Content; Organic Carbon Content; Nitrogen Content; Heartwood; Sapwood

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## 1. Introduction

The knowledge of organic carbon, ash and nitrogen levels in plants is very useful for the effective use of any plant. According to Meier et al. (2013), when wood is burned for fuel, nitrogen oxide (NO<sub>2</sub>) may be formed through the oxidation of nitrogen in the wood. Mitchual et al. (2014) indicated that nitrogen oxides (NOx) are beneficial since

they can enrich forest soils. However, negative impacts on health from the emission of high levels of NOx, including acidification of water and soils (Mitchual et al., 2014) and the risk of respiratory infections (Sillman, 2003) have been reported. Photochemical smog can be formed when NOx reacts with volatile organic compounds in the presence of sunlight and thus pollute the air (Sillman, 2003). Mitchual et al. (2014) indicated that carbon content contributes significantly to the heating value of wood fuel, with higher carbon content giving higher heating values.

According to Meier et al. (2013), carbon is the major constituent of wood, making up 45 to 50% of its biomass. The content of carbon, nitrogen and ash in plants is influenced by many factors. According to Kollmann (1959) and Campbell et al. (1990), factors that influence the amount and composition of ash include the season, weather conditions, and soil mineral availability.

Wood portions with higher densities result in an increase in the percentage of ash content as a result of the conversion of sapwood into heartwood from the bottom to the top portion of the tree, thus giving the bottom portion (with more heartwood and higher density) higher ash content than the top portion of the tree (Nurfaizah et al., 2014). Differences in carbon content between different parts of trees are small in relation to the range of variations in overall carbon content (Matthews, 1993). Ash content varies greatly within trees, highest at the heartwood and decreasing towards the bark (Imbeah, 1999). Mangroves are ecosystems that provide essential goods

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and services to human communities living in coastal areas. The benefits of mangroves include harvesting wood and non-wood products, fisheries, recreation, ecotourism, bio-filtration, coastal protection and carbon sequestration

(Spalding et al., 2010). This study aimed to determine the 55 organic carbon, nitrogen and ash content in Laguncularia Racemosa (White Mangroves) from the Central Region (CR) and Western Region (WR) of Ghana.

## 2. Materials and Method

## 2.1 Experimental

Matured laguncularia racemosa trees measuring 14-16 m in length and 30 cm in diameter were selected from Ghana's Central and Western regions. The felled Laguncularia racemosa (white mangrove) boles were divided into three sections: the base, middle and top.

## 2.2 Sample preparation

2 x 2 x 2 cm discs were cut from air-dried, defect-free billets prepared from each section of the trees from the two regions and labelled accordingly. Defect-free, airdried samples from each section were made into small chips and further milled with a high-speed laboratory blender and sieved with a 250 µm sieve.

## 2.3 Determination of ash content

- ASTM D 1102-84 (2007) standard was used to determine the ash content. An empty crucible was first ignited in a muffle furnace at 600°C, cooled in a desiccator, and weighed to the nearest 0.1 mg. 2 g air-dried wood samples were weighed into different pre-weighed crucibles and placed in an oven at  $103 \pm 2^{\circ}$ C, then, cooled in a
- desiccator and weighed. The heating and cooling were repeated until the weight was constant. The crucibles and contents were then placed in the muffle furnace at 600°C for 4 hours to burn off all the carbon. The sample was heated slowly at the start to avoid flaming while pro-
- tecting the crucibles from strong draughts at all times to 85 avoid mechanical loss of the test specimen. The temperature of the final ignition was 600°C. The crucibles, with their content, were then placed in a desiccator to cool and weigh. The heating was repeated until a constant weight of not more than 0.2 g was obtained after cooling.
- The ash content was calculated as in equation 1.

$$Ash(\%) = M_1 \times 100/M_2 \tag{1}$$

Where:  $M_1 = Weight ash M_2 = Weight of oven-dried$ sample.

### 2.4 Determination of organic carbon (C) content

The organic carbon content of the wood samples was determined by the Walkley – black wet oxidation method (Nelson & Sommers, 1982; Heanes, 1984). 0.1g samples from each of the three sections from the two regions were weighed into separate 500 mL Erlenmeyer flasks, followed by the addition of 10 mL of 1.0 N potassium 100 dichromate solution and 20 mL of concentrated H<sub>2</sub>SO4. Each sample mixture was swirled to ensure that the solution was in contact with all the particles of the wood samples. The flasks and their content were allowed to cool on an asbestos sheet for 30 minutes, after which 200 mL of distilled water and 10 mL of orthophosphoric acid were added. 2.0 mL (of 10 mL) of diphenylamine indicator was added, and the resulting solution was titrated with 1.0 N ferrous sulphate solution until the colour changed to blue and then finally to green. A blank determination 110 was made, and the carbon content of the samples was determined by the formula shown in equation 2.

$$\% OrganicC = Blank - (T \times N) \times 0.3 / Weight of sample$$
(2)

Where: Blank= Titre value for blank ( $\geq 10.5$ ). T= mL of FeSO<sub>4</sub> used for titration (titre value) N = Normalityof FeSO<sub>4</sub>

#### 2.5 Determination of nitrogen (N) content

The nitrogen content of the wood samples was determined by the Kjeldahl method (Bremner & Mulvaney, 1982). 2g air-dried samples from sections of the trees were separately weighed into different 500 mL long-necked Kjeldahl flasks, 120 and 10 mL distilled water was added to moisten the samples. A spatula full of Kjeldahl catalyst (mixture of 1-part selenium + 10 parts  $CuSO_4$  + 100 parts  $Na2SO_4$ ) was added, followed by 20 mL concentrated H2SO4. The solution was placed in a macro Kjeldahl digestion unit 125 to digest until the solution was clear and colourless. The flask was allowed to cool, the fluid was decanted into 100 mL volumetric flasks, and distilled water was added. An aliquot of 10 mL was transferred from the digested sample using a pipette into the Kjeldahl distillation flask, and 90 130 mL of distilled water was added to make up to 100 ml in the distillation flask. An additional 20 ml of 40% NaOH was added to the content of the distillation flask and distillate collected over 10 mL of 4% boric acid containing 3 drops of mixed indicator in a 200 mL conical flask. A 135 light blue colour was observed due to the presence of nitrogen. 100mL of distillate was titrated with 0.1 N HCl until the blue colour changed to grey and then to pink. A blank determination was carried out. The Nitrogen content  $(N_c)$  of the samples was determined using the 140 formula in equation 3.

$$\delta N_c = \frac{14 \times (A - B) \times N \times 100}{1000 \times 0.2}$$
 (3)

Where: A = volume of standard HCL used in sample titration B = volume of standard HCL used in blank titration N = normality of standard HCL

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$$\%N_c = \frac{2g \times 10mL}{100mL} = 0.2g \tag{4}$$

3. Results and Discussion

3.1 Evaluation of organic carbon content

Figure 1 shows the mean organic carbon content of Laquncularia racemosa samples from CR and WR.



Figure 1. Mean Organic Carbon Content of Laguncularia racemosa

BH = Heartwood of Base, BS = Sapwood of Base,MH = Heartwood of Middle, MS = Sapwood of Middle,150 TH = Heartwood of Top, TS = Sapwood of Top.

Figure 1 shows that WR samples recorded the greatest organic carbon content at the middle (31.92%) and lowest at its top (30.72%) along the sapwood. The heartwood recorded the highest at the base (31.59%), and the middle and top recorded the same (30.32%). Along the sapwood of trees from CR, the base recorded the greatest

value (31.92%) and lowest at its middle (30.32%). The heartwood recorded the greatest at the top (31.52%) and

least at both the base and the middle recorded the same

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(31.12%)(Figure 1). Figure 2 shows the mean carbon content along the

Stem (heartwood and sapwood portions) of Laguncularia racemose.

CB = Central Base, CM = Central Middle, CT = Cen-165 tral Top, WB = Western Base, WM = Western Middle, WT = Western Top. The overall organic carbon content along the stem (Figure 2) indicates the base recorded the highest value (31.52 and 31.56%, CR and WR, respectively) and lowest at the middle (30.72%) and top 170 (30.52%) for CR and WR respectively. There was a significant difference in mean carbon content (p < 0.05) for the various portions along the stem. According to Meier et al. (2013), carbon is the major constituent of wood, making up 45 to 50% of its biomass. From the study, the 175 organic carbon content (sapwood and heartwood) along the stem ranged from 30.72 - 31.52% for CR and 30.52- 31.56% for WR. According to Matthews (1993), differences in carbon content between different parts of trees



Figure 2. Mean Carbon Content along the Stem (heartwood and sapwood portions) of Laguncularia racemose

are small in relation to the range of variations in overall 180 carbon contents. The results for organic carbon content for L. racemosa in this research followed the same trend with relatively small differences between sapwood and heartwood for trees from each region.

#### 3.2 Evaluation of ash content

Figure 3 shows the mean ash content of Laguncularia racemosa samples from CR and WR.



Portions of Laguncularia racemosa

Figure 3. Mean Ash Content of Laguncularia racemosa

MS = Sapwood of Middle, TH = Heartwood of Top,TS = Sapwood of Top.

In Figure 3, CR samples had the highest ash content 190 at the base sapwood (3.29%), followed by the top sapwood (2.72%) and the middle sapwood (2.42%). The heartwood was similarly higher in ash content at the base portion (3.22%), followed by the top (2.72%) and the middle (2.35%) for the same region. WR samples had the highest mean value at the top for both sapwood and heartwood (2.98 and 2.89%, sapwood and heartwood, respectively) and the least at the middle portion for sapwood (2.47%)and at the base for heartwood (2.70%). The sapwood was generally higher in ash content than heartwood except 200 in the middle portion of samples from WR, where the heartwood was higher than the sapwood. Figure 4 shows

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the mean ash content along the stem (heartwood and sapwood portions) of Laguncularia racemose.



Figure 4. Mean ash content along the stem (heartwood and sapwood portions) of Laguncularia racemose

CB = Central Base, CM = Central Middle, CT = Cen-205 tral Top, WB = Western Base, WM = Western Middle, WT = Western Top.

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The ash content (heartwood and sapwood) along the stem (Figure 4) indicates that ash content was highest at the base (3.26%), followed by the top (2.72%) and the middle (2.39%) for CR, while it was highest at the top (2.94%) followed by the base (2.72%) and then the middle (2.62%) for WR. Although CR recorded a generally higher ash content at the base, it was comparatively low at the middle and top portions than WR. There was a significant difference in ash content (p < 0.05) for the different stem portions from the two regions.

According to Imbeah (1999), ash content varies greatly within trees, being highest at the heartwood and decreasing towards the bark. From this study, the heartwood 220 generally recorded a fairly higher ash content than the sapwood for both regions (Figure 3). The trees from CR had their base portions also recording a significantly higher ash content, whiles those from WR recorded a relatively higher ash content at the top portions (Fig-225 ure 4). However, a post hoc analysis indicated nearly no significant difference between the portions from WR and the corresponding portions along the heartwood and sapwood. According to Nurfaizah et al. (2014), wood portions with higher densities result in an increase in the 230 percentage of ash content as a result of the conversion of sapwood into heartwood from the bottom to the top

portion of the tree, thus giving the bottom portion (with more heartwood and higher density) higher ash content than the top portion of the tree. 235

The overall mean ash content recorded was 3.26 -2.39% for CR and 2.94 - 2.62% for WR (Figure 4). According to Ndlovu (2007), temperate-climate woods yield 0.1-1.0% ash, while tropical and sub-tropical woods yield

up to 5%. Campbell (1990) also indicates that, on average, the burning of wood results in about 6 - 10% ash. The ash content of L. racemosa from this study is within the expected range for tropical woods. According to Kollmann (1959) and Campbell (1990), other factors that influence the amount and composition of ash include the season, weather conditions, and soil mineral availability. CR samples with a generally higher ash content suggest that the soil from which the samples were harvested might have more minerals than those from WR. Thus, L. racemosa wood with a relatively low ash content may be useful as biofuel for industrial applications.

## 3.3 Evaluation of nitrogen (N) content

Figure 5 shows the mean nitrogen content of laguncularia racemosa from CR and WR.



Figure 5. Mean ash content along the stem (Mean nitrogen content of Laguncularia racemosa

BH = Heartwood of Base, BS = Sapwood of Base,255 MH = Heartwood of Middle, MS = Sapwood of Middle,TH = Heartwood of Top, TS = Sapwood of Top.

Nitrogen content was high at the top for sapwood (343.81 and 257.16 mg/Kg, CR and WR, respectively) and heartwood (256.82 and 228.17 mg/Kg, CR and WR 260 respectively) (Figure 5). However, CR samples recorded low values at the base for sapwood (237.82 mg/Kg) and middle for heartwood (234.31 mg/Kg), while WR samples recorded low values at the base (194.06 mg/Kg) for heartwood and middle for sapwood (204.90 mg/Kg) (Figure 265 5). Figure 6 shows the mean nitrogen content along the stem (heartwood and sapwood portions) of Laguncularia racemose from CR and WR.

CB = Central Base, CM = Central Middle, CT = Central Top, WB = Western Base, WM = Western Middle, 270 WT = Western Top.

From Figure 6, nitrogen content was generally high for CR, increasing from base to top along the stem (223.41 -300.32 mg/Kg compared to WR (199.55 – 242.66 mg/Kg). There was no significant difference in nitrogen content 275 except that of the base and middle portions of samples from WR. The combined nitrogen content (sapwood and heartwood) for L. racemosa ranged from 223.41 – 300.32 mg/Kg for CR and 199.55 - 242.66 mg/Kg for WR. According to Meier et al. (2013), when wood is burned for 280 fuel, nitrogen oxide  $(NO_2)$  may be formed through the

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**Figure 6.** Mean nitrogen content along the stem (combined heartwood and sapwood portions) of Laguncularia racemose

oxidation of nitrogen in the wood. Mitchual et al. (2014) indicated that levels of NOx could be beneficial since they enrich forest soil. Thus L. racemosa, from WR, with relatively low nitrogen content, is recommended for use as wood fuel since the amount of nitrogen oxides emission from its combustion can be relatively low. However, negative impact on health from the emission of high levels of NOx, including acidification of water and soils (Mitchual

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- et al., 2014) and the risk of respiratory infections (Sillman, 290 2003), has been reported. Sillman (2003) also indicated that photochemical smog could be formed when NOx reacts with volatile organic compounds in the presence of sunlight and thus pollute the air. Since the sapwood gen-
- erally recorded relatively high nitrogen content and the distribution along the stem largely increased from base to top (Figure 5), the sapwood and towards the top portions of L. racemosa may have the potential to pollute the environment and could have negative health effect when
- used as fuel. The values of nitrogen content recorded from 300 this study are, however, lower than what was reported by Mitchual et al. (2014) for A. robusta(481.3 mg/Kg), C. pentandra (481.7 mg/Kg) and T. scleroxylon (560.0 mg/Kg). The results from both regions were below the
- recommended levels (Nitrogen content  $\leq 0.3\%$ ) set by 305 the Austria national standard for pellets and briquettes, Austria ÖNORM M7135 (Mitchual et al., 2014). It is also below the German national standard for fuel pellets, except for the sapwood of the top portion for CR, Germany DIN 51731 /DINplus (Mitchual et al., 2014). This 310
- further confirms that L. racemosa, especially those from WR, may be ideal for use as wood fuel because of the relatively low nitrogen content.

## 4. Conclusion

The organic carbon, ash, and inorganic mineral elements composition of L. racemosa make it suitable as a biofuel energy source for industrial purposes. L. racemosa, from WR, has a relatively low nitrogen content compared to CR. It may be recommended for wood fuel since the amount of nitrogen oxide emission from its combustion 320 can be relatively low. The sapwood generally recorded relatively higher nitrogen content from L. racemosa.

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